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| NOIDA | Roll. No: INSTITUTE OF ENGINEERING AND TECHNOLOGY, GREATER NOIDA (An Autonomous Institute Affiliated to AKTU, Lucknow) B.Tech SEM: II - THEORY EXAMINATION - (2024- 2025) Subject: Engineering Physics |
| Time: 3 | |
| General In IMP: Verig 1. This Que Questions (2. Maximum) | Astructions: Ty that you have received the question paper with the correct course, code, branch etc. estion paper comprises of three Sections -A, B, & C. It consists of Multiple Choice (MCQ's) & Subjective type questions. The marks for each question are indicated on right -hand side of each question. The your answers with neat sketches wherever necessary. |
| | suitable data if necessary. |
| v | ly, write the answers in sequential order. |
| 6. No sheet evaluated/o | should be left blank. Any written material after a blank sheet will not be |
| ечанианеа/С | пескей. |
| SECTION | 7-A 20 |
| 1. Attempt | |
| • | What will be the rest energy of an electron? (CO1, K1) |
| (a) | 0.41 MeV |
| (b) | 0.51 MeV |
| (c) | 0.61 MeV |
| (d) | 0.71 MeV |
| 1-b. A | According to relativity, length of a rod in motion (CO1, K1) |
| (a) | Is same as its rest length |
| (b) | Is more than its rest length |
| (c) | Is less than its rest length |
| (d) | May be more or less than or equal to rest length depending on the speed of rod |
| 1-с. | The wavelength of the matter wave is independent of: (CO2, K1) |
| (a) | Mass |
| (b) | Velocity |
| (c) | Momentum |
| (d) | Charge |
| 1-d. | At what condition, $vp = vg? (CO2, K1)$ |
| (a) | $\frac{dv_p}{dk} = 0$ |
| (b) | |

| dv | | |
|-------------------------------|---------------------------------------------------------------------------|---|
| $\frac{d^{p}p}{d\lambda} = 0$ | | |
| dv " | | |
| (c) $\frac{P}{d\mu}$ | = 0 | |
| dv, | | |
| (d) $\frac{P}{d\omega}$ | = 0 | |
| | light wave suffers reflection at the interface between air and glass | 1 |
| | , the change of phase of the reflected wave in air is equal to: (CO3, K1) | |
| (a) 0 | | |
| (b) ∏ | | |
| (c) 2Π | | |
| (d) $\Pi/2$ | | |
| 1-f. The diff | raction Phenomenon is (CO3, K1) | 1 |
| (a) Bend | ling of light around an obstacle | |
| (b) Rect | ilinear propagation of light | |
| (c) Osci | llation of light wave in one direction | |
| (d) None | e of above | |
| 1-g. Number | lattice points in a primitive cell is (CO4, K1) | 1 |
| (a) One | Tattice points in a primitive cell is (CO4, K1) | |
| (b) Two | | |
| (c) Four | | |
| (d) Depe | ends on type of bravais lattice | |
| 1-h. Atomic | packing factor for BCC is (CO4, K1) | 1 |
| (a) 0.52 | | |
| (b) 0.74 | | |
| (c) 0.68 | | |
| | e of these | |
| 1-i. The tem (CO5, K | perature at which a conductor becomes a superconductor is known as (1) | 1 |
| (a) Curio | e temperature | |
| (b) Onno | e's temperature | |
| (c) Critic | cal temperature | |
| (d) None | e of these | |
| 1-j. The Che | emical bonding of carbon nano-tubes is composed entirely of (CO5, K1) | 1 |
| (a) sp bo | onds | |
| (b) sp2 b | oonds | |
| ` ′ 1 | | |
| · · · - | oonds | |

| 2. Atten | npt all parts:- | |
|---------------|----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|----|
| 2.a. | What do you understand by frame of reference? What are their types? (CO1, K1) | 2 |
| 2.b. | Distinguish between phase velocity and group velocity. (CO2, K1) | 2 |
| 2.c. | What is resolving power? (CO3, K1) | 2 |
| 2.d. | What do you understand by space lattice? CO4, K1) | 2 |
| 2.e. | What is the phenomenon of superconductivity? (CO5, K1) | 2 |
| SECTIO | <u>ON-B</u> | 30 |
| 3. Answ | er any five of the following:- | |
| 3-a. | At what speed will the mass of a body be 2.25 times its rest mass? (CO1, K3) | 6 |
| 3-b. | Calculate the amount of work to be done to increase the speed of an electron from 0.6c to 0.8c. Given that rest energy of electron = 0.5MeV. (CO1, K3) | 6 |
| 3-c. | Find the de-Broglie wavelength of a neutron of energy 12.8 MeV. (CO2, K3) | 6 |
| 3-d. | Calculate the energy difference between the ground state and first excited state for electron in one dimensional rigid box of length 10 ⁻⁸ cm. (CO2, K3) | 6 |
| 3.e. | Find the minimum number of lines in a plane diffraction grating required to just resolve the sodium doublet (5890 & 5896 Angstrom) in the first order and second order. (CO3, K3) | 6 |
| 3.f. | Calculate the inter planner spacing for (1 0 1) plane in a simple cubic crystal whose lattice constant is 0.42 nm. (CO4, K3) | 6 |
| 3.g. | Explain Temperature dependence of resistivity in superconductors. (CO5, K2) | 6 |
| SECTIO | <u>DN-C</u> | 50 |
| 4. Answ | er any <u>one</u> of the following:- | |
| 4-a. | Deduce the relativistic velocity addition theorem. Show that it is consistent with Einstein's second postulate of special theory of relativity. (CO1, K2) | 10 |
| 4-b. | Describe Michelson -morely experiment and explain the outcome of the experiment. (CO1, K2) | 10 |
| 5. Answ | er any <u>one</u> of the following:- | |
| 5-a. | What is uncertainty principle? How will you explain non existence of electrons in the nucleus? (CO2, K2) | 10 |
| 5-b. | Derive the time independent and time dependent Schrodinger wave equations. (CO2, K2) | 10 |
| 6. Answ | er any <u>one</u> of the following:- | |
| 6-a. | Explain the experimental arrangement of Newtons ring. Why the center of newtons ring is dark? Find the refractive index of a liquid with the help of newtons ring formation. (CO3, K2) | 10 |
| 6-b. | Give the construction and theory of plane transmission grating and explain the formation of spectra by it. (CO3, K2) | 10 |
| 7. Answ | rer any <u>one</u> of the following:- | |
| 7-a. | Describe the structure of NaCl crystal, CO4, K2) | 10 |

| /-D. | what are the different types of Bravais lattice in the 7 crystal system. CO4, K1) | 10 |
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| 8. Answe | er any <u>one</u> of the following:- | |
| 8-a. | Explain Meissner effect. Show that superconductors become perfect diamagnetic in an external magnetic field. (CO5, K2) | 10 |
| 8-b. | Describe C_{60} buckyballs. Give some properties and uses of Buckyballs. (CO5, K2) | 10 |

